

Fullerene in metal-organic framework: design of a novel material

Michel Côté

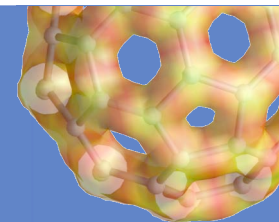
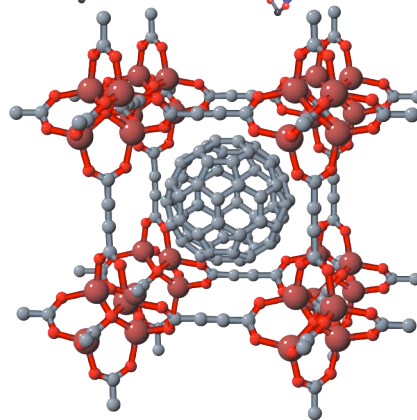
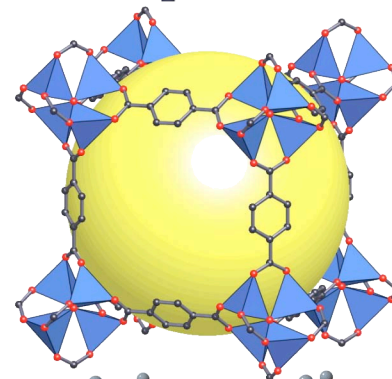
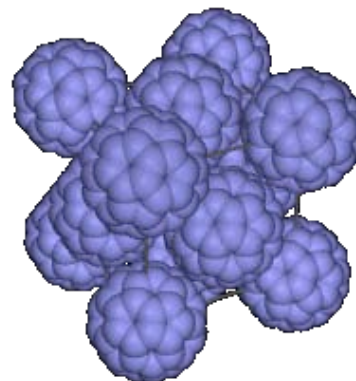
Département de physique

Regroupement québécois sur les matériaux de pointe (RQMP)

**Université 
de Montréal**

Outline:

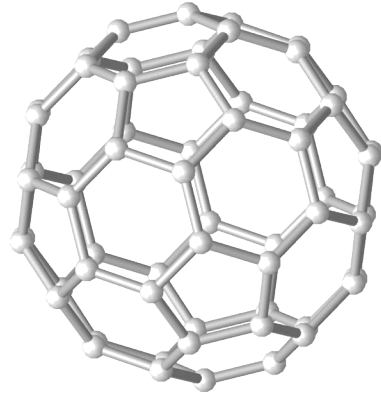
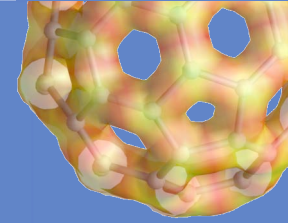
- C_{60} and its crystal
 - C_{60} properties
 - Superconductivity in C_{60} solid
- Metal-organic-framework (MOF)
 - Structure
 - Applications
- A new compound: C_{60} and MOF
 - Design of electronic structure



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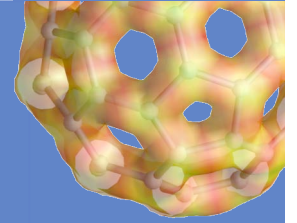
C₆₀ fullerene



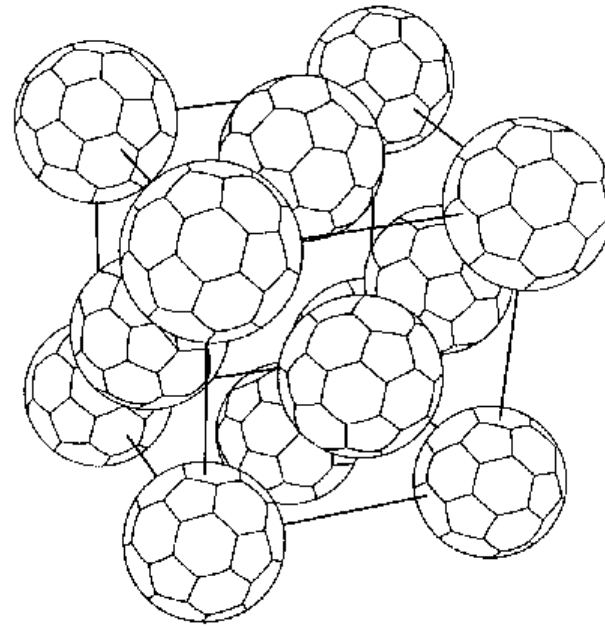
Biosphere in Montreal

- name, *fullerenes* ou *buckyballs*, is in honor of the architect Buckminster Fuller
- discover in 1985; Nobel Prize in 1996
Curl, Kroto and Smalley

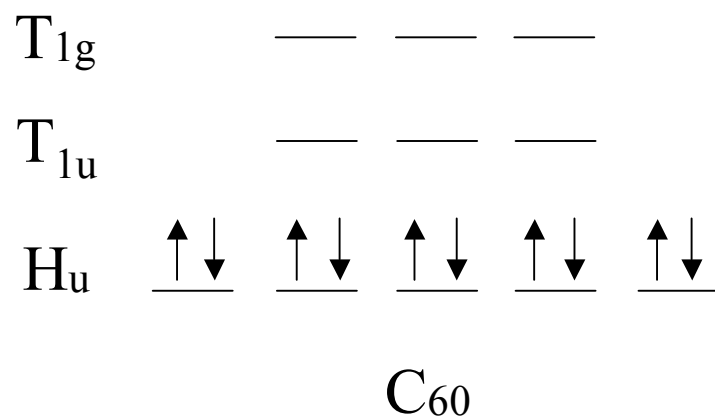
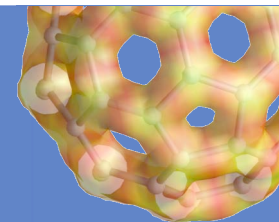
C₆₀ properties



- high symmetry; icosahedral
- formation energy = 8.4 eV/atom
For comparison graphite/diamond = 8.8 eV/atom
- very stable
Can be thrown at 100 000 km/h
at a wall without breaking.
- can form a solid

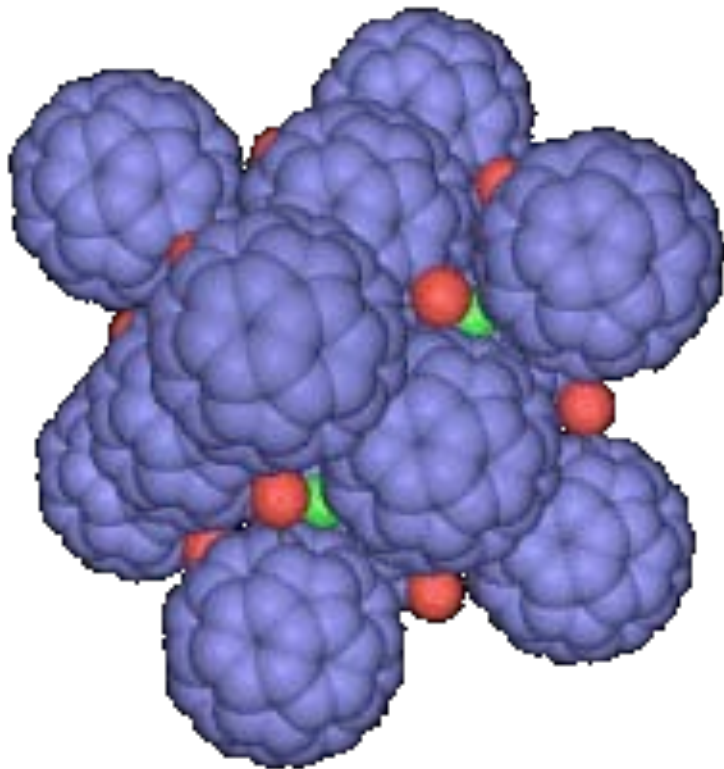
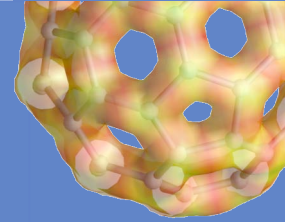


Electronic structure of C_{60}



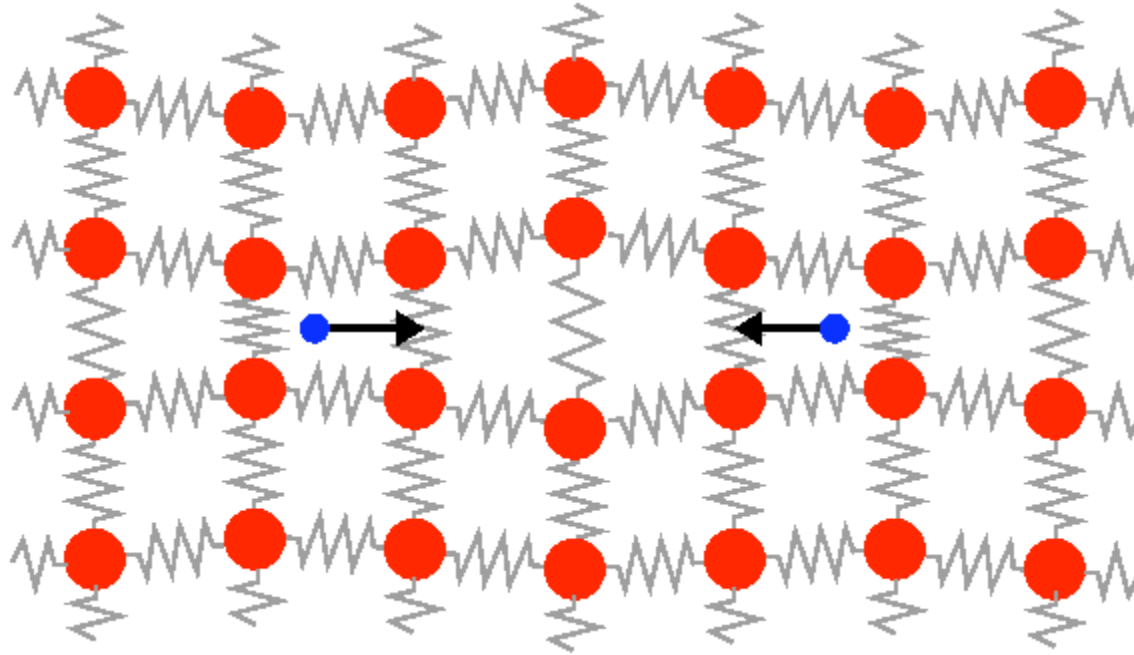
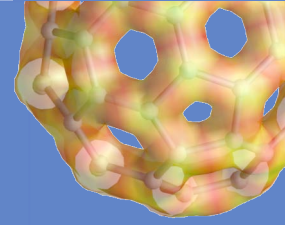
In order to make C_{60} a metal with a high electronic density of states, we want to have 3 electrons for each C_{60} , hence A_3C_{60} , where A is an alkali atom.

C₆₀ metal



- C₆₀ crystals can be doped with alkali.
A₃C₆₀ (A=K,Rb,Cs).
- T_c ~ 10 - 40 K linked to the enlargement of the distance between the C₆₀ (10 to 10.3 Å center to center).

Superconductivity



- effective attraction between electrons due to motion of atoms.

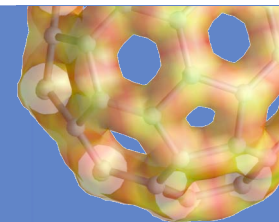
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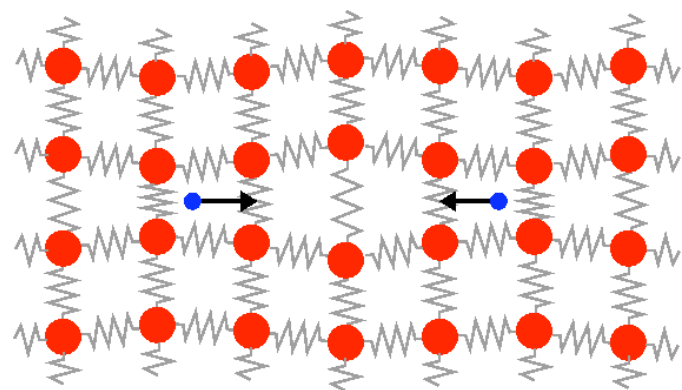
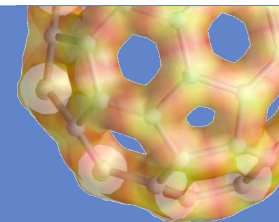
Transition Temperature

$$T_C \sim \theta_D e^{-1/V_{elph} D(\epsilon_F)}$$

- The interaction responsible for the superconductivity in fullerenes is mediated via the intramolecular phonons

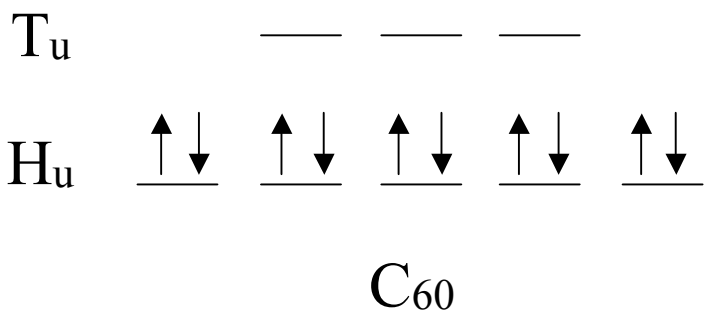


Electron-phonon coupling

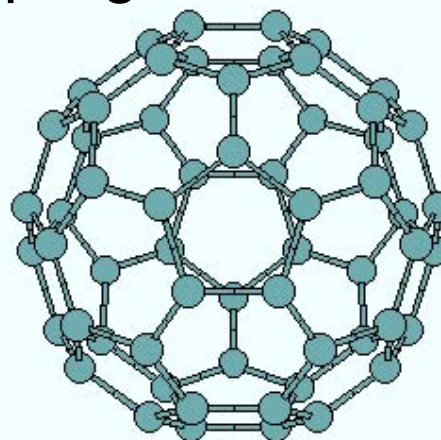


Only intramolecular modes are responsible for the coupling

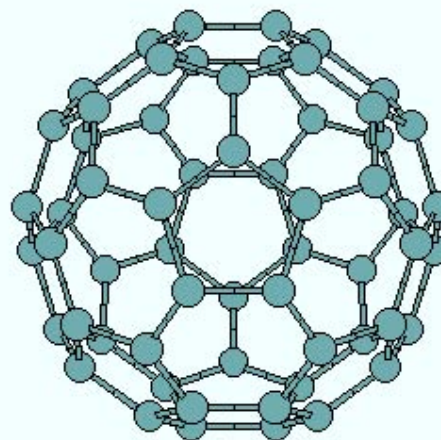
$$V_{ep} \sim \sum_{\lambda} \frac{|\langle \psi_i | \epsilon_{\lambda} \cdot \delta V | \psi_j \rangle|^2}{\omega_{\lambda}^2}$$



Ag mode



Hg mode

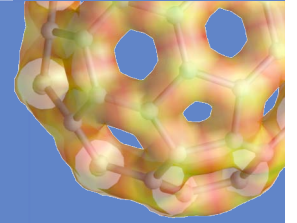


Transition Temperature

$$T_C \sim \theta_D e^{-1/V_{elph} D(\varepsilon_F)}$$

- The interaction responsible for the superconductivity in fullerenes is mediated via the intramolecular phonons
- When the lattice parameter changes, only the **DOS at the Fermi level** changes

$$D(\varepsilon_F) \sim \frac{1}{\text{Dispersion}} \sim \text{Lattice parameter}$$



T_c vs lattice parameter

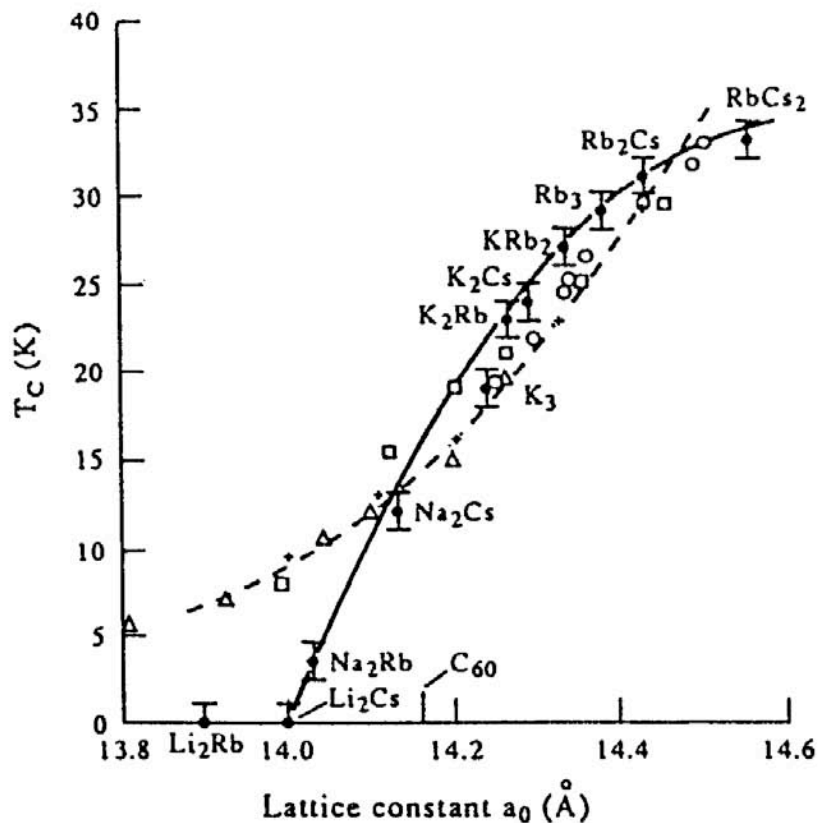
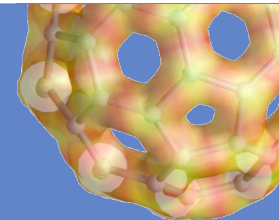
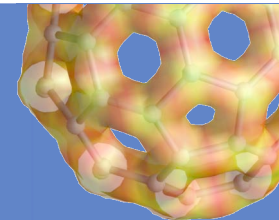


Fig. 10.14. The relationship between T_c and the lattice parameter a_0 for A_3C_{60} (A: Li, Na, K, Rb, Cs and their binary alloys) superconductors. Open triangles and squares are from pressure experiments, and the dotted line represents the T_c - a_0 relationship expected from the simple BCS theory using the density of states due to LDA calculations. The solid line is a guide for the eyes. From [10.48]

Source: T. Ishiguro, K. Yamaji, G. Saito, "Organic Super-conductors"

Alkali atoms



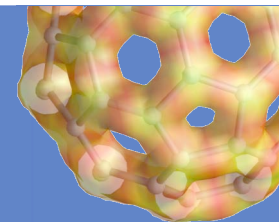
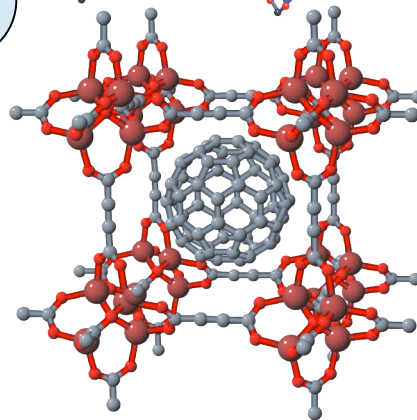
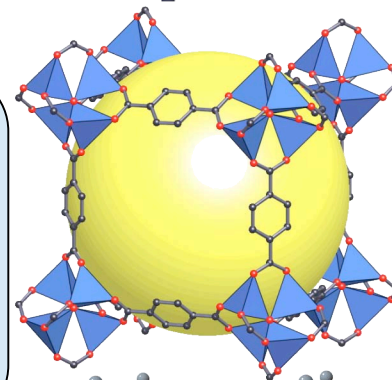
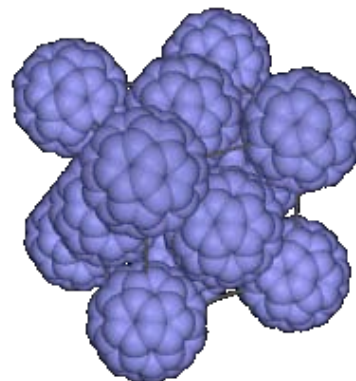
	IA				
1	1 H	IIA			
2	3 Li	4 Be			Colors
3	11 Na	12 Mg	IIIB	IVB	VB
4	19 K	20 Ca	21 Sc	22 Ti	23 V
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb
6	55 Cs	56 Ba	57 La	72 Hf	73 Ta
7	87 Fr	88 Ra	89 Ac	104 Rf	105 Ha

$T_c = 18\text{ K}$ ←
 $T_c = 28\text{ K}$ ←
 $T_c = 40\text{ K}$ ←
 $T_c = ?$ ←

Fr half-life = 22 minutes

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Hydrogen storage

Fuel cell:

⇒ the problem of hydrogen storage

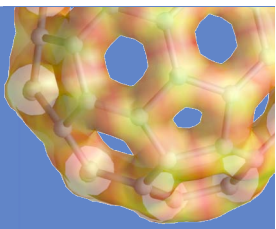
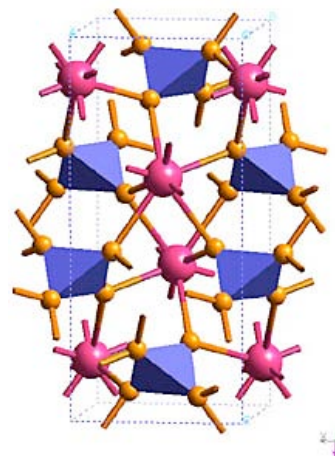
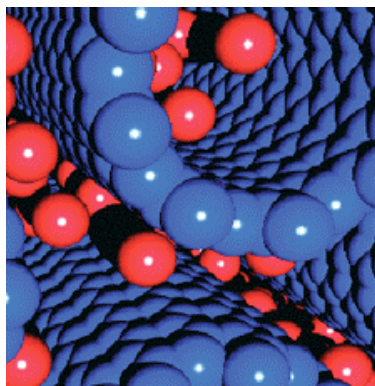
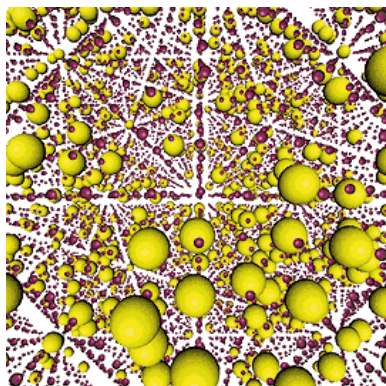
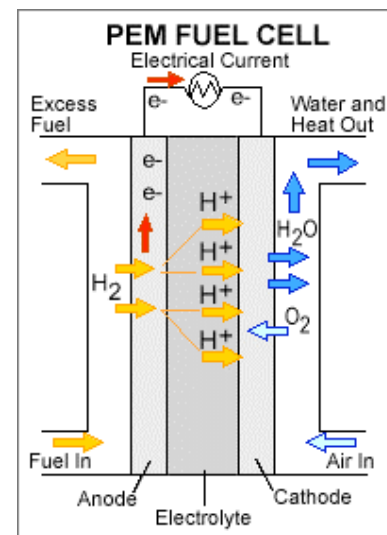
Present material capacity:

⇒ 2 to 4 % of their weight in hydrogen

DOE goal:

⇒ 6.5 % of their weight

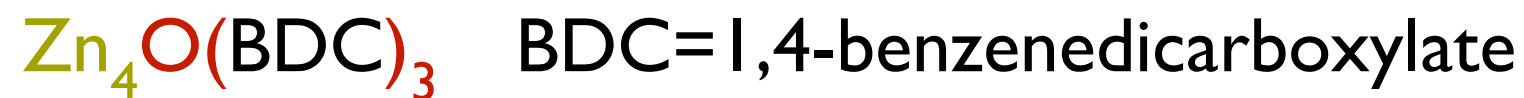
⇒ 1.2 billion US\$ in research funding



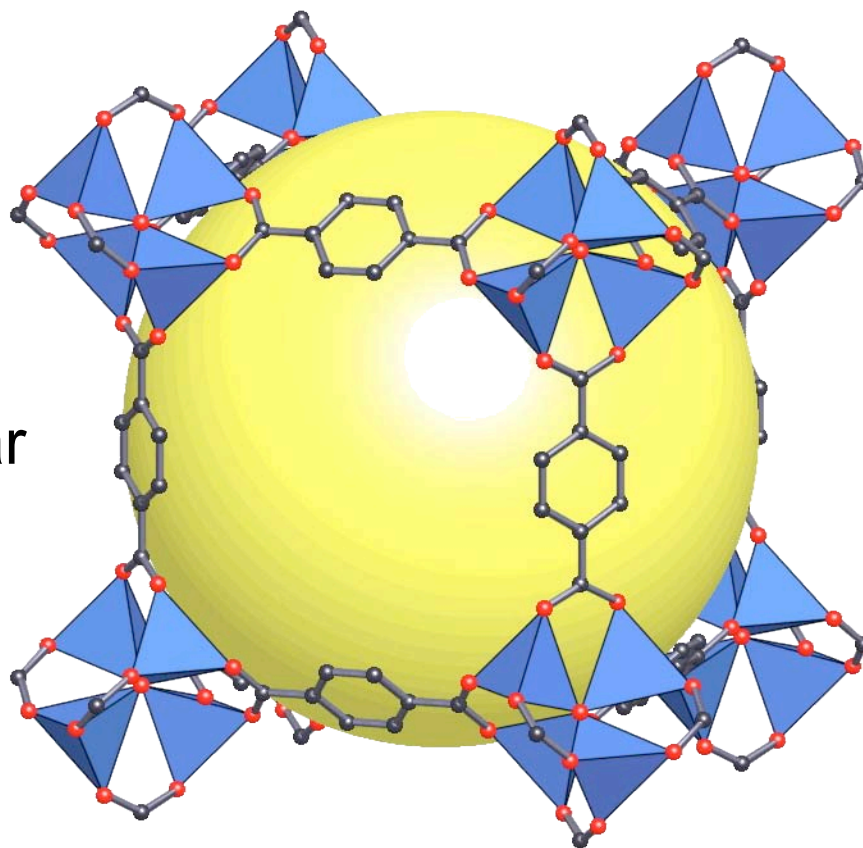
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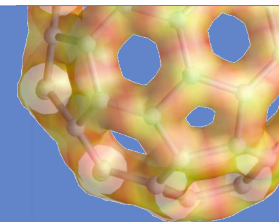
Metal-Organic Framework (MOF)



- 4.5 weight % of H_2 at 78 K and 0.8 bar
- 0.5% at RT and 10 bar
- 1% at RT and 20 bar



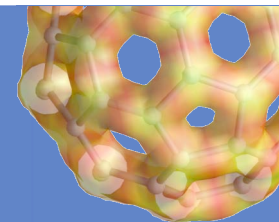
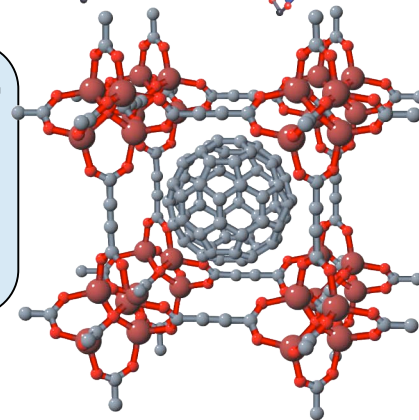
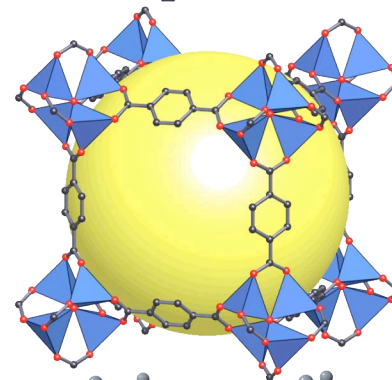
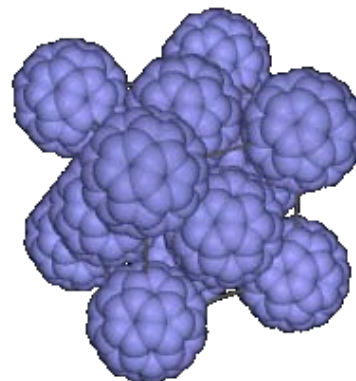
Rosi et al., Science **300** (2003)
Yaghi's group, Chemistry, Michigan University.



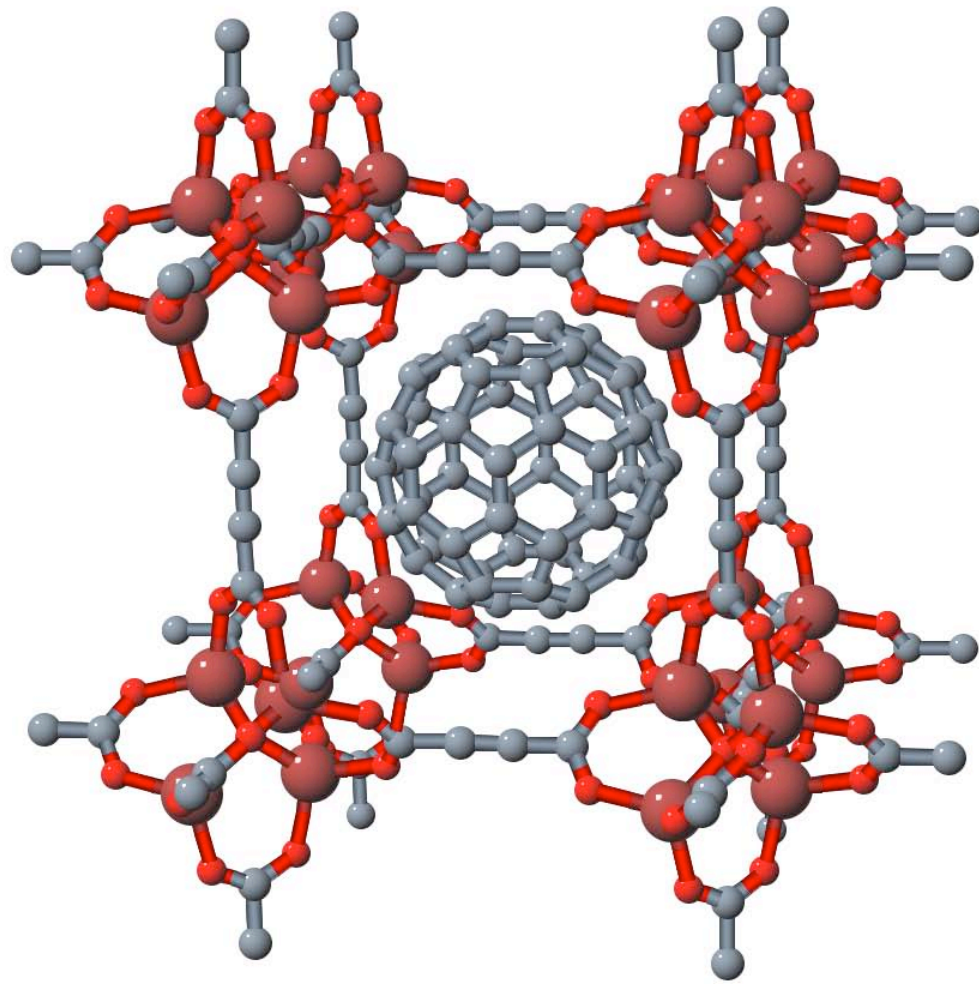
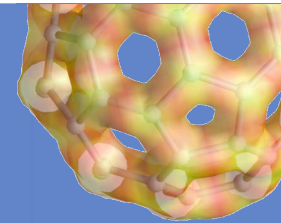
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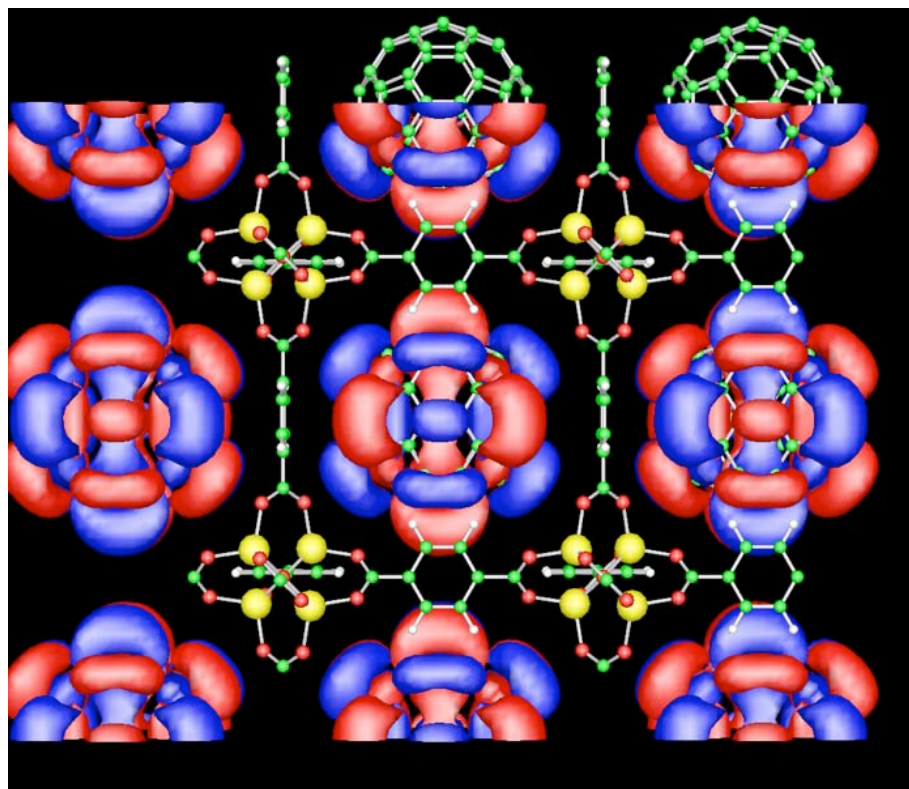
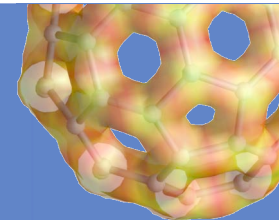


Material Design

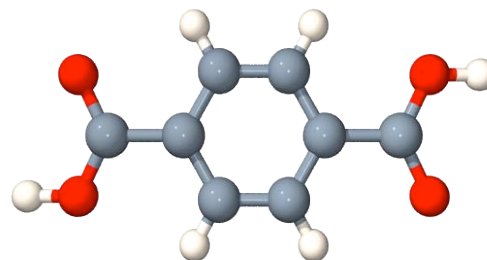


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MOF-5C₆₀



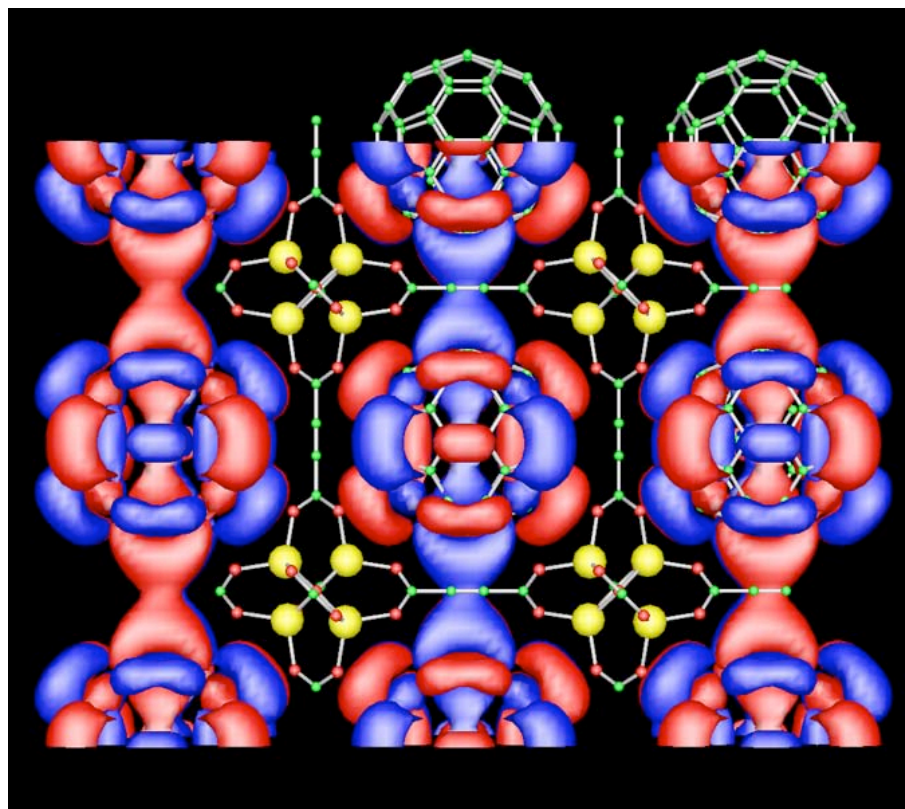
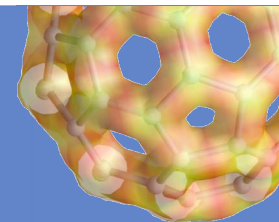
- $\Delta E = 0.87 \text{ eV}$
- New cubic structure for C₆₀
- Center to center distance of 12.9 Å between the C₆₀
- Need to reduce the size of the organic linker 1,4-benzenedicarboxylic



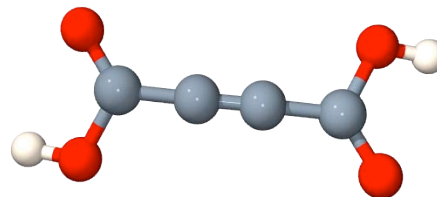
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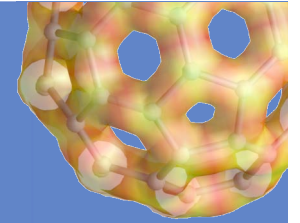
- $\Delta E = 1.02 \text{ eV}$
- Keep the simple cubic structure for the C_{60}
- Center to center distance of 11.3 \AA between the C_{60}
- The organic acid 2-butynediolic in the synthesis of the MOF



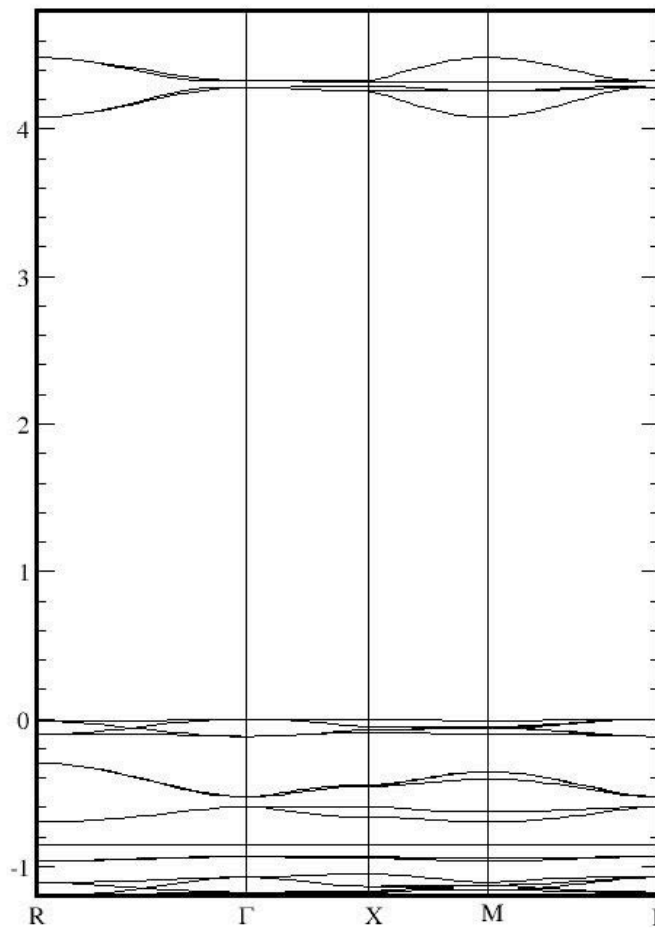
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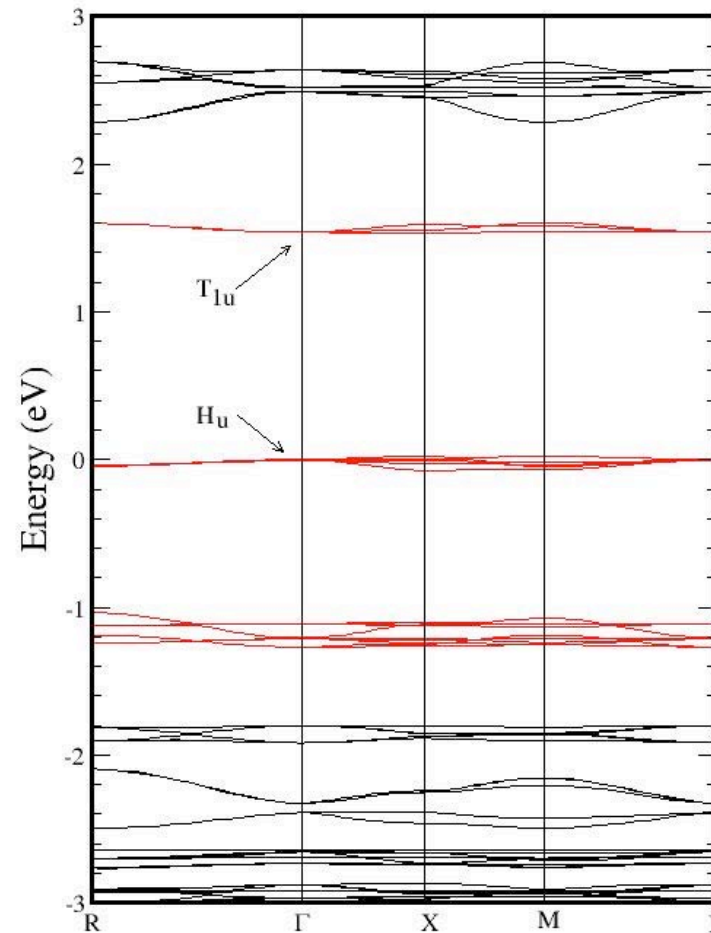
MOFC₄C₆₀ band structure



MOF - Simple Cubic (C = 11.42 Ang)



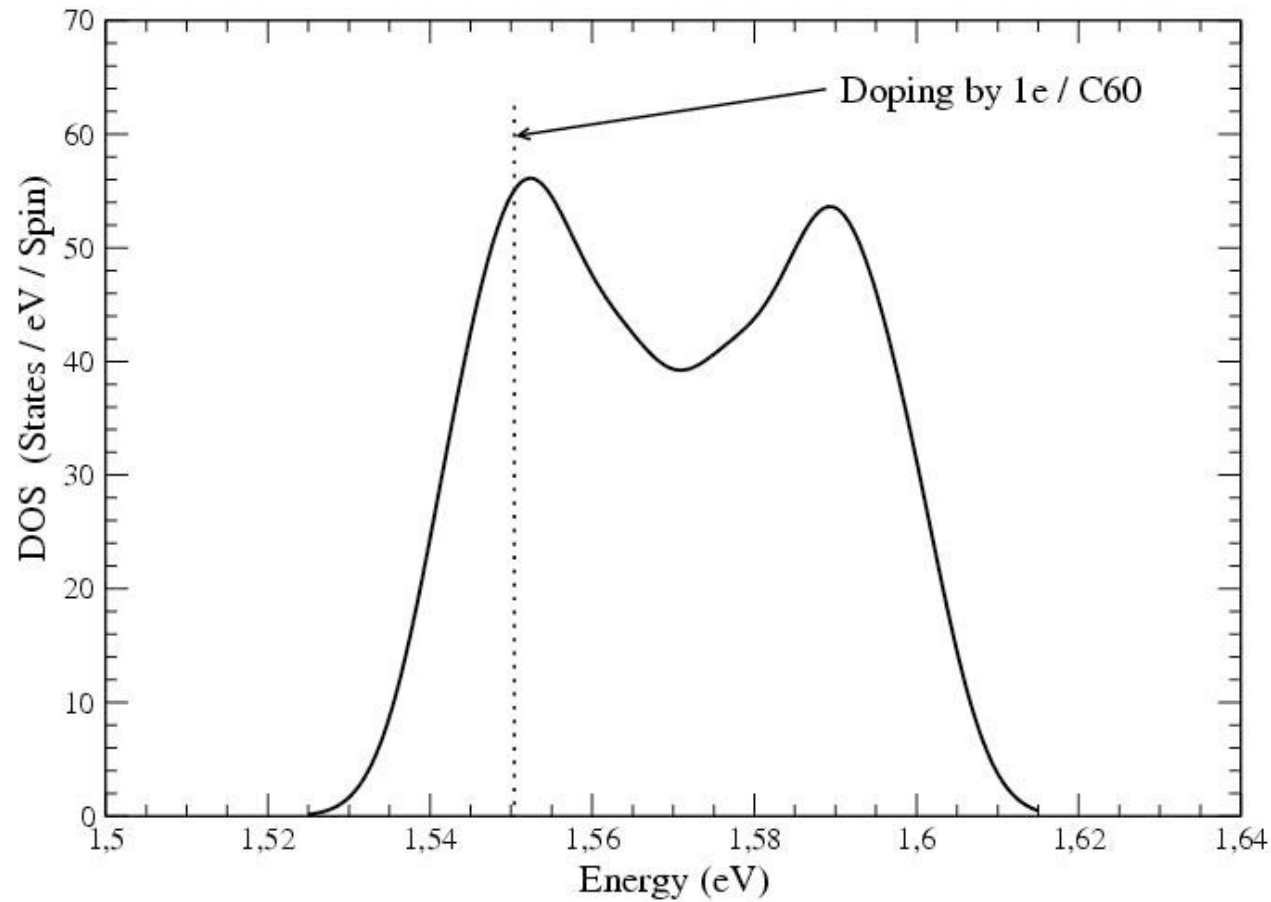
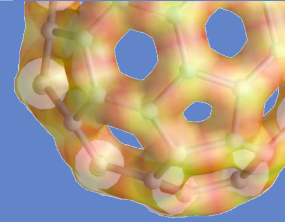
C60@MOF - Simple Cubic (C = 11.42 Ang)



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Density of states

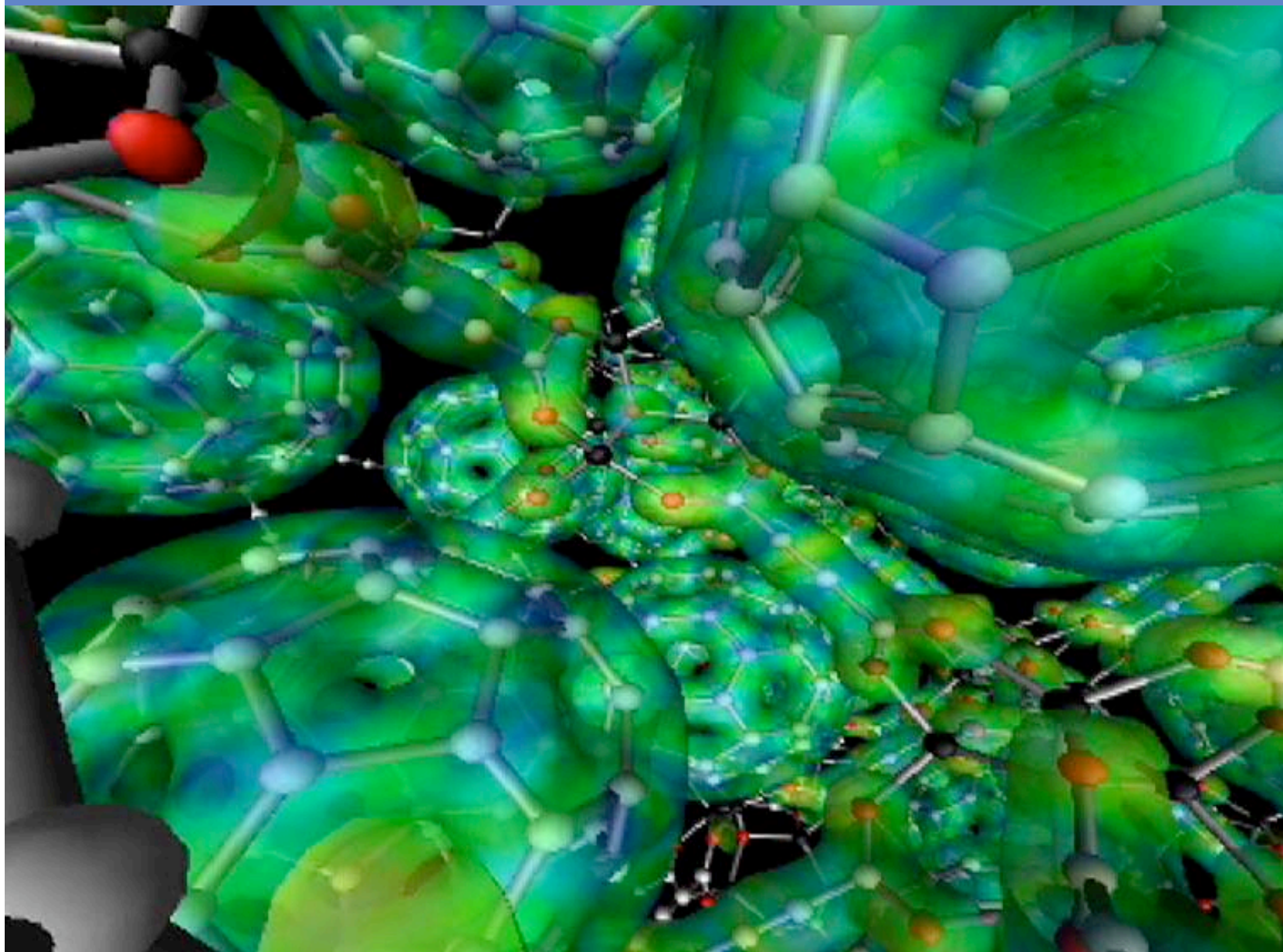
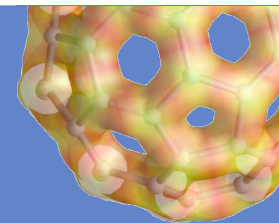


Solid FCC C₆₀: 15 states/eV/spin

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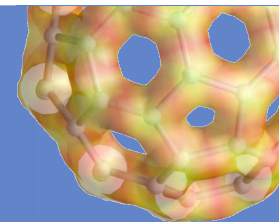
A look at the charge density...



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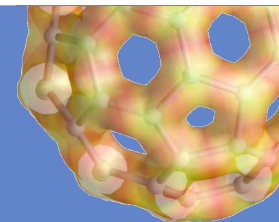
Conclusion:



- Electronic properties can be tailor with the combinaison of MOF and C60.
- Many new possibilities remain to be explore.
- The synthesis has started.

For more details see: PRL (95) 146403 (2005)

Members of the group



Postdocs who worked on this project:

- Vladimir Timochevskii (McGill)
- Sébastien Hamel (Livermore)



Present graduate students:

- Paul Boulanger
- Guillaume Dumont
- Sébastien Langevin
- Simon Pesant
- Benjamin Tardif



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